

# Fluoride-Catalyzed Addition of PhSCF<sub>2</sub>SiMe<sub>3</sub> to *N*-Substituted Cyclic Imides Followed by Radical Cyclization: General Synthetic Strategy of *gem*-Difluoromethylenated 1-Azabicyclic Compounds

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PhSCF<sub>2</sub>SiMe<sub>3</sub> (1) was found, for the first time, to undergo fluoride-catalyzed nucleophilic difluoro-(phenylsulfanyl)methylation reaction to cyclic imides **2**, affording the corresponding adducts **3** in moderate to good yields. Reductive cleavage of the phenylsulfanyl group of *N*-alkylated adducts **3** with Bu<sub>3</sub>SnH/ AIBN yielded *gem*-difluoromethylated products **4**. Under the same reduction conditions, *N*-alkenylated and *N*-alkynylated adducts **3** afforded the corresponding *gem*-difluoromethylenated 1-azabicyclic compounds **5** and **6** with *trans* stereoselectivity. These compounds were employed as precursors for preparing substituted *gem*-difluoromethylenated pyrrolizidinones and indolizidinones **7** and **8** by treatment with Et<sub>3</sub>SiH/BF<sub>3</sub>•OEt<sub>2</sub>, and compounds **9** and **10** by nucleophilic displacement of the hydroxyl group, using organosilanes in the presence of BF<sub>3</sub>•OEt<sub>2</sub>. The synthesis of highly substituted *gem*-difluoromethylenated pyrrolizidines **13** and **14** was also demonstrated.

#### Introduction

*gem*-Difluoromethylene moiety has been incorporated into bioactive compounds due to its properties as an isopolar and isosteric substituent for an oxygen.<sup>1</sup> The presence of the *gem*-difluoromethylene group can enhance the biological properties

of drug molecules.<sup>2</sup> Therefore, syntheses of *gem*-difluoromethylenated analogs of natural products have been extensively investigated.<sup>3</sup> Few general synthetic methods for constructing

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gem-difluoromethylenated compounds were previously reported.<sup>4</sup> It is still desirable to develop new flexible and facile strategies for the preparation of these compounds. Recently, PhSCF<sub>2</sub>SiMe<sub>3</sub> (1) has been demonstrated as a useful gem-difluoromethylene building block.<sup>5</sup> Compound 1 reacts extensively with carbonyl derivatives,<sup>6</sup>  $\gamma$ -ketoesters,<sup>7</sup> imines,<sup>8</sup> and alkyl bromides,<sup>9</sup> providing convenient routes for the preparation of the corresponding gem-difluoromethylated alcohols,  $\gamma$ -lactones, amines, and alkanes, respectively. Furthermore, the stereoselective difluoromethylenation of chiral imines using PhSCF<sub>2</sub>SiMe<sub>3</sub> (1) and the synthesis of chiral disubstituted gem-difluoromethylenated pyrrolidines have been recently reported.<sup>10a</sup> In addition, alternative new entries to assemble various fluorinated *S*-, *O*-, and *N*-heterocycles have also been developed.<sup>10b-d</sup>

As a part of our ongoing research interest in developing methodologies for the preparation of 1-azabicyclic derivatives,<sup>11</sup> we sought for a general method to synthesize *gem*-difluoromethylenated 1-azabicyclic analogs, employing **1** as a *gem*difluoromethylene building block. It has been demonstrated that cyclic imides can undergo fluoride-catalyzed trifluoromethylation, using (trifluoromethyl)trimethylsilane (CF<sub>3</sub>SiMe<sub>3</sub>).<sup>12</sup> Hence, it is anticipated that fluoride-catalyzed nucleophilic difluoro-

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SCHEME 1. Fluoride-Catalyzed Nucleophilic Difluoro(phenylsulfanyl)methylation of 1 with Imides 2 and Synthetic Conversion of the Resulting Adducts 3 into Compounds 4, 5, and 6 and Subsequently to Compounds 13 and 14



(phenylsulfanyl)methylation of 1 to cyclic imides 2 would generate the corresponding adducts 3. When R groups of adducts 3 are allylic and homoallylic, reductive cleavage of the corresponding phenylsulfanyl moiety could afford radical intermediate that should then undergo intramolecular radical cyclization, providing *gem*-difluoromethylenated 1-azabicyclic compounds 5 and 6, respectively (Scheme 1). Furthermore, nucleophilic displacement of the hydroxyl group and subsequent organometallic addition to carbonyl group followed by reduction would lead to highly substituted difluoromethylenated 1-azabicyclic compounds 13 and 14. The proposed synthetic operations are summarized in Scheme 1.

### **Results and Discussion**

The initial investigation involved fluoride-catalyzed nucleophilic difluoro(phenylsulfanyl)methylation of imide 2a with PhSCF<sub>2</sub>SiMe<sub>3</sub> (1). The expected adduct **3a** was obtained in 95% yield when the reaction was carried out using 10 mol % of anhydrous tetrabutylammonium fluoride (TBAF) in dry THF at -78 °C to room temperature overnight followed by acidic workup (2 M HCl; Table 1, entry 1). To demonstrate the generality of the reaction, the reactions of 1 with N-substituted phthalimide and succinimide derivatives, which were prepared by base-catalyzed N-alkylation of imides<sup>13</sup> with alkylating agents (Table 1, entries 1-6, 8, 10-13, 15 and 17), Mitsunobu reaction of imides with alcohols<sup>14</sup> (Table 1, entries 7, 14 and 16), and cross-olefin metathesis<sup>15</sup> of N-alkenylated imides with alkenes (Table 1, entries 9 and 18), were investigated. The results are summarized in Table 1. Except for some cases, the reactions of 1 with imide derivatives conducted under the standard conditions as for **3a** readily gave the corresponding adducts; yields range from moderate to good (Table 1). In case of imide 2i, under the standard conditions, adduct with double bond transposition was obtained as a major product (33% yield). Gratifyingly, the reaction employing tetrabutylammonium triphenyldifluorosilicate (TBAT) in place of TBAF gave adduct 3i in moderate yield

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entry	imides 2	$3 (\% \text{ yield})^{a,b}$	4, 5 or 6	entry	imides 2	$3 (\% \text{ yield})^{a,b}$	4, 5 or 6
entry	initides 2	5 (70 yield)	(% yield <sup>a</sup> , trans : cis)			5 (70 yield)	(% yield <sup>a</sup> , trans : cis)
1	N-CH <sub>2</sub> Ph	$Ho F SPh N-CH_2Ph 3a (95)$	$HO F H HO HO CH_2Ph$	10	2j	HOF SPh	<b>5e</b> (81, 90:10) <sup>6</sup>
2	2h	HOF = HOF = SPh $HOF = SPh$		11		HO N N N	HO F CH <sub>2</sub> Ph
3		$\frac{HO}{N-COOEt}$	$4\mathbf{b} (70)$	12	2k	$3\mathbf{k} (68)$ $HOF F SPh$ $N - $ $3\mathbf{l} (83)$	5f (87, 66:34) <sup>5</sup> Ho 5g (85, 64:36) <sup>6</sup>
4	A-CH <sub>2</sub> Ph	HOF F N-CH <sub>2</sub> Ph	HOF FH N-CH <sub>2</sub> Ph dt (66)	13		HOF SPh SIMe3	
5	L N-CH <sub>3</sub> 2e	$HO F SPh N-CH_3 3e (68)$	$4\mathbf{u}(00)$	14	2m SiMe <sub>3</sub>	$\frac{3m (68)^{c}}{N}$	Sh (73, E:Z=80:20) Ho $F$ SiMe <sub>3</sub> Si (60, E:Z=85:15)
6		HO SPh 3f(69)	<b>5a</b> (73, 94:6) <sup>x</sup>	15	20	HOFFSPh N 30 (50) <sup>c</sup>	HONN Sj (45, 50:50) <sup>g</sup>
7	C C C C C C C C C C C C C C C C C C C	HOFFSPhPh 3g (68)	<b>5b</b> (73, 77:23) <sup>8</sup>	16	ر بر المراجع من المراجع 2p	3p (76)	<b>6a</b> (63, 77:23) <sup>e</sup>
8		<b>3h</b> (90, <i>E</i> : <i>Z</i> =79:21)	<b>5c</b> (58, 62:38) <sup>2</sup>	17	2q	HOFF SPh	<b>6b</b> (63, 72:28) <sup>4</sup>
9	2i		HO N Ed (5) 53:40 %	18		HOFF SPh Ph	

TABLE 1.	Preparation of Adducts 3	3. gem-Difluoromethylated	Lactams 4. and	gem-Difluoromethylenated	1-Azabicvclic Com	ounds 5 or 6
		.,		a		

<sup>*a*</sup> Isolated yields. <sup>*b*</sup> Unless otherwise noted, all reactions were carried out employing 10 mol % TBAF/THF, -78 °C to rt. <sup>*c*</sup> By using 10 mol % TBAT/DMF, -78 °C to rt. <sup>*d*</sup> By using 10 mol % TBAT/THF, -78 to 15 °C, 5 h. <sup>*e*</sup> By using 10 mol % TBAT/THF, -78 °C to rt. <sup>*f*</sup> Ratio of *E*- and *Z*-isomers was determined by <sup>1</sup>H NMR integration of the crude product. <sup>*g*</sup> Ratios of isomers (*trans/cis*) were determined by <sup>1</sup>H NMR integration of the crude product.

(55%) and the double bond transposition adduct was not observed (Table 1, entry 9). It is also worth mentioning that the reaction of 1 with *N*-trimethylsilylpropargyl succinimide 2n under the standard conditions furnished the expected product 3n in only 2% yield, but the desilylated adduct 3o was obtained as a major product (30% yield). However, the reaction employing 10 mol % of TBAT in dry DMF at -78 °C to room temperature gave the desired product 3n in 45% yield together with 3o (10% yield; Table 1, entry 14). Similar results, that is, 3n (40% yield) and 3o (8% yield) were obtained when the

reaction was carried out using THF as the solvent. Interestingly, treatment of 1 with *N*-propargylphthalimide 2m employing 10 mol % TBAT in DMF at -78 °C to room temperature unexpectedly provided trimethylsilylated derivative 3m in 68% yield (Table 1, entry 13).

The reductive desulfenylation of adducts 3a-e was achieved by treating with Bu<sub>3</sub>SnH and a catalytic amount of AIBN in refluxing toluene. The *gem*-difluoromethylated products 4a-ewere obtained in moderate to good yields (66–86%; Table 1,

SCHEME 2. Proposed Transition States for the Radical Cyclization

# 

(major)

entries 1–5).<sup>6b,16</sup> The *gem*-difluoromethyl radical intermediate, generated by CF<sub>2</sub>–SPh bond cleavage of the adducts **3** containing the *N*-unsaturated substituents, could be trapped intramolecularly by an unsaturated functional moiety,<sup>17</sup> affording *gem*-difluoromethylenated 1-azabicyclic compounds **5**. Thus, under similar radical conditions, intramolecular radical cyclization of **3f** readily proceeded to provide **5a** in 73% yield as a 94:6 mixture of *trans*- to *cis*-isomers together with a small amount of the corresponding reduced product. A single *trans*-**5a** was obtained in pure form after purification by either preparative thin-layer chromatography on silica gel or crystallization. Its relative stereochemistry was established by X-ray crystallography (see Supporting Information).

To demonstrate the efficiency and viability of our methodology, series of *gem*-difluoromethylenated pyrrolizidine derivatives 5b-g, as mixtures of isomers, were synthesized, in moderate to good yields, by treatment of adducts 3g-1 under standard radical conditions (Table 1, entries 7–12). Based on the X-ray crystallographic data of the major *trans*-isomer of 5a, we

assumed that the *trans*-isomers of **5b**-**g** were the major isomers. Notably, similar radical cyclization of N-propargyl substituted adducts proceeded smoothly, providing the cyclized products in moderate yields (Table 1, entries 13-15). Cyclization of silvlated alkynyl derivatives 3m and 3n afforded compounds 5h and 5i as a mixture of E- and Z-isomers. The E-isomer was obtained as a major isomer, of which the stereochemistry was established by NOE experiments (see Supporting Information). Irradiation of the olefinic proton of 5h or 5i led to no enhancement of the allylic protons. The formation of tributylstannylated product 5j (Table 1, entry 15) presumably resulted from consecutive addition of the tributylstannyl group to the triple bond followed by gem-difluoromethyl radical addition to the resulting vinylstannane derivative.<sup>18</sup> The presence of the exocyclic methylene moiety in compounds of types 5h and 5i should be valuable for further synthetic modification.

(minor)

Having established an efficient access to *gem*-difluoromethylenated pyrrolizidine derivatives **5**, we further demonstrated that our method could be used as a general route for the synthesis of *gem*-difluoromethylenated indolizidine derivatives **6**. Thus, *N*-homoallylated adducts **3p**-**r** were treated with Bu<sub>3</sub>SnH/AIBN in refluxing toluene, yielding the corresponding *gem*-difluoromethylenated indolizidines **6a**-**c** as a mixture of *trans*- and *cis*-isomers (Table 1, entries 16–18). The relative stereochemistry of the major *trans*-**6a** was confirmed by X-ray crystallography (see Supporting Information).

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SCHEME 3. Lewis Acid-Catalyzed Nucleophilic Substitution of the Hydroxyl Group of Compounds 5b and 6a



The stereochemical outcomes of compounds **5** and **6** can be rationalized as shown in Scheme 2. The radical mediated cyclization was proposed to proceed via 5-*exo* or 6-*exo* cyclization mode.<sup>19</sup> Transition states **A** and **C** are energetically more favorable due to minimized steric interaction between the pseudoaxial hydroxyl group and the vinylic double bond, resulting in the formation of thermodynamically more stable *trans*-isomer of pyrrolidine or piperidine heterocycles.

Having a general route to synthesize derivatives 5 and 6, we subsequently focused on the synthetic utilities of these adducts as precursors for the preparation of substituted gem-difluoromethylenated pyrrolizidines and indolizidines. It is anticipated that the presence of hydroxyl group in adducts 5 and 6 should provide a convenient access to an iminium intermediate, 20,21 which, in principle, can be trapped by an appropriate nucleophile. The results are summarized in Scheme 3. The adducts 5b (trans/cis; 77:23) and 6a (trans/cis; 88:12) were reduced with BF3•OEt2 and Et3SiH22 in CH2Cl2 at -78 °C to room temperature, generating the corresponding lactams 7 (77% yield) and 8 (98% yield) as a single isomer, as confirmed by the NOE experiments (see Supporting Information). Alkylation of 5b with Grignard reagents (MeMgBr or *i*-PrMgCl in THF in the presence of  $BF_3 \cdot OEt_2$  at 0 °C to rt, overnight) or organocoppers (n-Bu<sub>2</sub>CuLi or n-Bu<sub>2</sub>Cu(CN)Li<sub>2</sub> in THF in the presence of BF<sub>3</sub>•OEt<sub>2</sub> at -30 °C to rt overnight) was unsuccessful, and only

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SCHEME 4. Proposed Transition States for the Formation of *trans*-Isomers of Products 7–10



starting material was recovered. However, allylation reaction of **5b**, employing allyltrimethylsilane with BF<sub>3</sub>•OEt<sub>2</sub> in CH<sub>2</sub>Cl<sub>2</sub> at -78 °C to rt overnight, led to 9a in low yield (10% yield) together with recovery of 5b. Gratifyingly, when the reaction was performed under refluxing conditions for 3 h, 9a was obtained in moderate yield (67% yield). Similarly, under the same conditions, 10a was synthesized in 53% yield from the corresponding compound 6a. Both compounds 9a and 10a were obtained as a single isomer (see Supporting Information). Similarly, treatment of **5b** and **6a** with cinnamyltrimethylsilane afforded adducts 9b and 10b, respectively, as a mixture of diastereomers, as shown in Scheme 3. The reactions of 5b and 6a with phenyltrimethylsilylacetylene employing trimethylsilyl triflate (TMSOTf) as the Lewis acid at -78 °C to rt overnight provided the respective adducts 9c (54% yield as a 93:7 mixture of isomers) and 10c (67% yield as a 86:14 mixture of isomers). The use of BF3 • OEt2, SnCl4, and TiCl4 did not furnish to the desired products but led to the recovery of the starting material **5b** and phenylacetylene. This result may be due to extensive decomposition of phenyltrimethylsilylacetylene to phenylacetylene prior to its addition to iminium ion intermediate. The relative trans stereochemistries of adducts 7 and 9 could be rationalized based on the fact that the nucleophiles (hydride or organosilanes) attack the initially formed iminium ion intermediate from the face opposite to the benzyl group to avoid steric interaction (Scheme 4). In addition, attack of nucleophiles to the iminium ion derived from 6a from the pseudoaxial direction

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SCHEME 5. Preparation of Alkyl-Substituted gem-Difluoromethylenated Pyrrolizidines 11, 13, and 14 and Indolizidine 12



could be further explained by stereoelectronic principle, leading preferably to the *trans*-isomers of adducts **8** and **10**.

The conversions of compounds 9a and 10a to alkylsubstituted gem-difluoromethylenated pyrrolizidines and indolizidines proceeded smoothly (Scheme 5). Reduction of compounds 9a and 10a using LiAlH<sub>4</sub> in refluxing THF provided the corresponding gem-difluoromethylenated pyrrolizidine 11 (84% yield) and indolizidine 12 (60% yield), respectively (Scheme 5). The reaction of **9a** with n-BuLi/CeCl<sub>3</sub> in THF<sup>23</sup> followed by acidification generating in situ iminium ion intermediate, which was subsequently reduced by sodium cyanoborohydride (NaBH<sub>3</sub>CN) to afford butyl-substituted gemdifluoromethylenated pyrrolizidine 13 in 80% yield as a single diastereomer. Similarly, a single diastereomer of isopropylsubstituted gem-difluoromethylenated pyrrolizidine 14 was produced in 65% yield. The relative stereochemistries of compounds 11-14 were established by NOE experiments (see Supporting Information). The observed stereochemical outcomes of the reactions can be explained that the addition of organometallic reagents followed by acidification afforded iminium ions which were preferably reduced from the less sterically hindered face opposite to the allyl substituent (Scheme 5). It is worth mentioning that compound 10a did not react with either n-BuLi/CeCl<sub>3</sub> or *i*-PrMgCl in THF under the same conditions as for 9a.

## Conclusion

In conclusion, we have successfully developed for the novel fluoride-catalyzed nucleophilic difluoro(phenylsulfanyl)methylation reactions of PhSCF<sub>2</sub>SiMe<sub>3</sub> to succinic and phthalimide derivatives. The resulting adducts were employed as useful precursors for the preparation of *gem*-difluoromethylenated pyrrolizidines and indolizidines. The synthetic operation involves sequential intramolecular radical cyclization, nucleophilic displacement of the hydroxyl functionality using Lewis acid/ allyltrimethylsilanes or trimethylsilylacetylenes, organometallic addition employing *n*-BuLi/CeCl<sub>3</sub> and *i*-PrMgCl onto the carbonyl group of the  $\gamma$ -lactam moiety, followed by reduction of the resulting adducts with NaBH<sub>3</sub>CN. The synthetic application of our method for preparation of *gem*-difluoromethylenated analogs of some polyhydroxylated natural products is currently being investigated.

### **Experimental Section**

General Procedure for the Preparation of Compounds 3. 2-Benzyl-3-(difluoro(phenylsulfanyl)methyl)-3-hydroxy-isoindolin-1-one (3a). To a mixture of compound 1 (0.928 g, 4 mmol) and 2-benzylisoindoline-1,3-dione (2a; 0.470 g, 2 mmol) in THF (5 mL), was added 10 mol % TBAF (0.4 mL, 0.4 mmol, 1 M solution in THF). The reaction mixture was stirred at -78 °C followed by slowly warming up to room temperature overnight. The solution was quenched with 1 M HCl (3 mL) and extracted with EtOAc (3  $\times$  25 mL). The organic phase was washed successively with water and brine and dried over anhydrous Na2SO4. After solvent removal, the crude product was purified by radial chromatography (SiO<sub>2</sub>, 10-20% EtOAc in hexanes) to give a white crystal of 3a (0.753 g, 95% yield, mp = 142-144 °C). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 7.86-7.83 (m, 1H, ArH), 7.72-7.68 (m, 1H, ArH), 7.60-7.58 (m, 2H, ArH), 7.35-7.25 (m, 3H, ArH), 7.20-7.10 (m, 7H, ArH), 4.95 (d, J = 15.6 Hz, 1H, CHH), 4.45 (d, J = 15.6 Hz, 1H, CHH), 3.85 (br s, 1H, OH). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 168.3 (C=O), 141.9 (C), 137.5 (2 × C), 136.7 (2 × CH), 132.8 (CH), 131.5 (C), 131.0 (2 × CH), 130.1 (CH), 129.24 (t, J = 187.5 Hz, CF<sub>2</sub>), 129.0 (2 × CH), 128.5 (3 × CH), 127.4 (CH), 124.2 (CH), 123.8 (CH), 91.2 (t, J = 25.0 Hz, C), 43.5 (CH<sub>2</sub>). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>):  $\delta$  -80.48 (d, J = 211.0 Hz, 1F), -81.57 (d, J = 211.0 Hz, 1F). IR (nujol): v<sub>max</sub> 3207br, 1682s, 1612s, 1612m, 1497m, 1471s, 1456s, 1455s, 1427m, 1397s, 1361s, 1163s, 1137m, 1108m, 1073m, 1061s, 1027m, 1012m, 974m, 945m, 907m, 897m, 886m, 845m, 823m, 770s, 752s, 709s, 700s, 690s cm<sup>-1</sup>. MS: *m/z* (%) relative intensity 398 (M<sup>+</sup>, 10), 380 (7), 250 (6), 238 (47), 161 (11), 160 (100), 91 (20), 65 (7). HRMS (ESI-TOF) Calcd for  $C_{22}H_{17}F_2NO_2SNa$  [M + Na]<sup>+</sup>, 420.0846; found, 420.0851.

General Procedure for the Reductive Cyclization of Compounds 3 to Compounds 5. Preparation of 1,1-Diffuoro-9bhydroxy-2-methyl-2,3-dihydro-1*H*-pyrrolo[2,1-a]isoindol-5(9b*H*)one (5a). Argon was bubbled through a solution of 3f (0.374 g, 1 mmol) in toluene (2 mL) for 30 min, and a mixture of Bu<sub>3</sub>SnH (0.47 mL, 1.75 mmol) and AIBN (25 mg, 0.15 mmol) in toluene

<sup>(23)</sup> Nukui, S.; Sodeoka, M.; Sasai, H.; Shibasaki, M. J. Org. Chem. 1995, 60, 398–404.

(8 mL) was added dropwise at reflux over a 1 h period followed by refluxing for an additional 4 h. Volatiles were evaporated and the tin byproduct were removed by column chromatography (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub> then EtOAc) to give a crude product, which was purified by column chromatography (SiO<sub>2</sub>, 20% EtOAc in hexanes) to afford white crystals of **5a** (0.191 g, 73% yield, mp = 122-125 °C) as a 94:6 mixture of trans- and cis-isomers. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, cis-isomer marked\*): & 7.68-7.64 (m, 6H, ArH of trans- and cisisomers), 7.57-7.55 (m, 2H, ArH of trans- and cis-isomers), 4.12\* (dd, J = 12.0, 9.3 Hz, 1H, CHH), 3.72 (br s, 2H, OH of trans- and cis-isomers), 3.68-3.63 (m, 1H, CHH), 3.39-3.27 (m, 3H, CHH of the trans-isomer and CH of trans- and cis-isomers), 3.13\* (dd, *J* = 12.0, 4.9 Hz, 1H, CHH), 1.50\* (d, *J* = 7.5 Hz, 3H, CH<sub>3</sub>), 1.19 (d, J = 6.8 Hz, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  170.7 (C=O), 141.4 (C), 133.2 (CH), 132.9 (C), 130.9 (CH), 123.9 (CH), 123.2 (CH), 122.9 (t, J = 258.2 Hz, CF<sub>2</sub>), 93.1 (t, J = 25.1 Hz, C), 46.4 (d, J = 7.4 Hz, CH<sub>2</sub>), 39.3 (t, J = 22.7 Hz, CH), 9.5 (d, J = 7.1 Hz, CH<sub>3</sub>). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>, *cis*-isomer marked\*):  $\delta$  -101.16\* (dd, J = 232.2, 23.5 Hz, 1F), -126.86\* (d, J = 232.2 Hz, 1F), -127.82 (dd, J = 220.9, 24.9 Hz, 1F), -129.60(dd, J = 220.9, 6.8 Hz, 1F). IR (nujol):  $\nu_{\text{max}}$  3237br, 1681s, 1616s, 1469s, 1378m, 1239m, 1147m, 1123m, 1958m, 1003m, 759s, 709m, 701m, 604m cm<sup>-1</sup>. MS: *m/z* (%) relative intensity 239 (M<sup>+</sup>, 33) 203 (22), 201 (62), 200 (100), 188 (66), 172 (23), 161 (34), 160 (42), 145 (27), 133 (48), 132 (22), 125 (25), 117 (59), 91 (28), 77 (41). HRMS (ESI-TOF) Calcd for  $C_{12}H_{11}F_2NO_2Na$  [M + Na]<sup>+</sup>, 262.0656; found, 262.0634.

General Procedure for the Preparation of Compounds 9 and 10. Preparation of (2R\*,9bR\*)-9b-Allyl-2-benzyl-1,1-diffuoro-2,3dihydro-1H-pyrrolo[2,1-a]isoindol-5(9bH)-one (9a). BF<sub>3</sub>·OEt<sub>2</sub> (0.6 mL, 4.93 mmol) and allyltrimethylsilane (2.6 mL, 16.5 mmol) were added to a stirred solution of 5b (1.04 g, 3.3 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (12 mL) at room temperature and the mixture was heated at reflux for 3 h. The reaction was quenched with a saturated aqueous sodium hydrogen carbonate and extracted with  $CH_2Cl_2$  (3 × 20 mL). The combined extracts were washed with brine and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Filtration followed by evaporation gave a crude product, which was purified by preparative thin-layer chromatography (SiO<sub>2</sub>, 10% EtOAc in hexanes) to give a white solid of 9a as a transisomer (0.75 g, 67% yield, mp = 91–93 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.83 (d, J = 7.6 Hz, 1H, ArH), 7.63 (td, J = 7.5, 1.2 Hz, 1H, ArH), 7.54 (td, J = 7.5, 1.0 Hz, 1H, ArH), 7.50 (d, J = 7.6 Hz, 1H, ArH), 7.35 (t, J = 7.6 Hz, 2H, ArH), 7.30–7.25 (m, 3H, ArH), 5.31 (ddt, J = 17.1, 10.1, 7.1 Hz, 1H, CH), 5.03 (dd, J = 17.1, 1.4 Hz, 1H, CHH), 4.97 (d, J = 10.0 Hz, 1H, CHH), 3.73 (dd, J = 12.1, 8.7 Hz, 1H, CHH), 3.56 (dd, J = 12.1, 9.5 Hz, 1H, CHH), 3.39-3.29 (m, 1H, CH), 3.11 (dd, J = 13.9, 5.3 Hz, 1H, CHH), 2.91 (dd, J = 14.1, 7.0 Hz, 1H, CHH), 2.75 (dd, J = 13.9, 9.4 Hz, 2H, CHH). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 172.1 (C=O) 142.8 (d, J = 4.3 Hz, C), 137.9 (C), 133.8 (C), 132.4 (CH), 129.7 (CH), 129.4 (CH), 128.74 (2 × CH), 128.67 (2 × CH), 126.8 (CH), 124.9 (dd, J = 263.4, 255.4 Hz, CF<sub>2</sub>), 124.2 (CH), 122.5 (CH), 120.2 (CH<sub>2</sub>), 74.5 (t, J = 25.3 Hz, C), 48.3 (t, J = 22.6 Hz, CH), 45.1 (d, J = 7.4 Hz, CH<sub>2</sub>), 37.0 (d, J = 2.4 Hz, CH<sub>2</sub>), 32.9 (d, J = 7.8 Hz, CH<sub>2</sub>). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>):  $\delta$  -118.14 (dd, J = 226.5, 23.0 Hz, 1F), -121.05 (dd, *J* = 226.5, 7.1 Hz, 1F). IR (KBr):  $\nu_{\rm max}$  1703s, 1643w, 1612w, 1491w, 1467m, 1363s, 1220s, 1049s, 751m, 704s cm<sup>-1</sup>. MS: m/z (%) relative intensity 340 (M<sup>+</sup> + 1, 11), 339 (M<sup>+</sup>, 2), 298 (100), 249 (20), 91 (64), 65 (11). HRMS (ESI-TOF) Calcd for  $C_{21}H_{19}F_2$ NONa [M + Na]<sup>+</sup>, 362.1332; found, 362.1319.

Preparation of  $(2R^*,9bR^*)$ -9b-Allyl-2-benzyl-1,1-difluoro-2,3,5,9btetrahydro-1*H*-pyrrolo[2,1-*a*]isoindole (11). To a suspension of LiAlH<sub>4</sub> (4.3 mg, 1.13 mmol) in THF (3 mL) was added a solution of **9a** (0.153 g, 0.45 mmol) in THF (4 mL). The mixture was heated at reflux overnight and quenched at 0 °C by careful addition of water (0.5 mL) followed by 1 M NaOH (0.5 mL). The resulting mixture was filtered through a Celite pad and the filtrate was concentrated under reduced pressure. Purification of the crude product by column chromatography (SiO2, 5% EtOAc in hexanes) gave a pale yellow oil of 11 (0.123 g, 84% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.30-7.25 (m, 5H, ArH), 7.21-7.15 (m, 4H, ArH), 5.50 (ddt, J = 17.2, 10.1, 7.0 Hz, 1H, CH), 4.95 (dd, J = 17.2, 1.8 Hz, 1H, CHH), 4.91 (d, J = 10.2 Hz, 1H, CHH), 4.23 (d, *J* = 15.1 Hz, 1H, CHH), 3.76 (d, *J* = 15.1 Hz, 1H, CHH), 3.29 (t, J = 7.4 Hz, 1H, CHH), 3.02 (dd, J = 13.9, 5.1 Hz, 1H, CHH), 2.90-2.86 (m, 1H, CH), 2.63-2.58 (m, 2H, CHH, CHH), 2.46 (dd, J = 13.9, 7.4 Hz, 1H, CHH), 2.42 (ddd, J = 11.2, 9.1, 1.7 Hz, 1H, CHH). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 140.7 (C), 139.1 (C), 138.8 (d, J = 7.5 Hz, C), 132.8 (d, J = 1.4 Hz, CH), 128.6 (2 × CH), 128.5 (2 × CH), 127.9 (CH), 127.0 (CH), 126.9 (t, J = 261.0 Hz, CF<sub>2</sub>), 126.3 (CH), 123.9 (CH), 122.5 (CH), 118.4 (CH<sub>2</sub>), 81.5 (t, J = 22.8 Hz, C), 60.2 (CH<sub>2</sub>), 57.3 (d, J = 9.1 Hz, CH<sub>2</sub>), 46.6 (t, J = 21.6 Hz, CH), 43.1 (t, J = 3.0 Hz, CH<sub>2</sub>), 31.0 (d, J = 6.0 Hz, CH<sub>2</sub>). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>):  $\delta$  -109.23 (dd, J = 224.2, 23.0 Hz, 1F), -115.08 (dd, J = 223.7, 6.1 Hz, 1F). IR (neat):  $\nu_{\text{max}}$ 1641m, 1456s, 1218s, 724s cm<sup>-1</sup>. MS: *m/z* (%) relative intensity 278 (9), 236 (100), 216 (7), 194 (18), 176 (7), 130 (6). HRMS (ESI-TOF) Calcd for  $C_{21}H_{22}F_2N$  [M + H]<sup>+</sup>, 326.1722; found, 326.1767.

Preparation of (2R\*,5S\*,9bR\*)-9b-Allyl-2-benzyl-5-butyl-1,1difluoro-2,3,5,9b-tetrahydro-1H-pyrrolo[2,1-a]isoindole (13). To a solution of CeCl<sub>3</sub> (0.65 g, 1.75 mmol) in THF (5 mL) was added n-BuLi (1.2 mL, 1.74 mmol, 1.46 M solution in hexane) at -78 °C. After stirring at -78 °C for 1 h, a solution of 9a (0.12 g, 0.35 mmol) in THF (5 mL) was added. The mixture was stirred at -78  $^{\circ}$ C for 10 h and then gradually warmed to -20  $^{\circ}$ C, and quenched by the addition of 4 M HCl-dioxane in MeOH (4 M HCl-dioxane/ MeOH, 1:29, 5 mL) followed by excess NaBH<sub>3</sub>CN. The resulting reaction mixture was stirred at 0 °C for 1 h and 10% aqueous NaOH solution (5 mL) was then added. The mixture was diluted with  $CH_2Cl_2$  and extracted with  $CH_2Cl_2$  (3 × 20 mL). The combined extracts were washed with brine and dried over anhydrous Na2SO4. Filtration followed by evaporation gave a crude product, which was purified by radial chromatography (SiO<sub>2</sub>, 5% EtOAc in hexanes) to afford a colorless oil of **13** (0.094 g, 80% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.31-7.18 (m, 8H, ArH), 7.13-7.11 (m, 1H, ArH), 5.59 (ddt, J = 17.3, 10.4, 7.1 Hz, 1H, CH), 4.96-4.92 (m, 2H, CH<sub>2</sub>), 3.83 (dd, *J* = 9.1, 3.4 Hz, 1H, CH), 3.40 (t, *J* = 7.9 Hz, 1H, CHH), 3.04 (dd, J = 13.7, 4.9 Hz, 1H, CHH), 2.97–2.90 (m, 1H, CH), 2.63 (dd, J = 13.7, 9.7 Hz, 1H, CHH), 2.57–2.53 (m, 2H, CHH, CHH), 2.47 (dd, J = 13.9, 7.3 Hz, 1H, CHH), 1.72–1.68 (m, 1H, CHH), 1.52-1.26 (m, 5H, CHH,  $2 \times CH_2$ ), 0.89 (t, J =7.3 Hz, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 144.9 (C), 139.2 (C), 138.2 (d, J = 6.8 Hz, C), 133.9 (CH), 128.7 (2 × CH), 128.5  $(2 \times CH)$ , 127.9 (CH), 127.0 (dd, J = 263.5, 251.9 Hz, CF<sub>2</sub>), 127.0 (CH), 126.3 (CH), 123.0 (CH), 122.3 (CH), 118.0 (CH<sub>2</sub>), 80.5 (t, J = 22.0 Hz, C), 73.5 (CH), 58.4 (d, J = 9.6 Hz, CH<sub>2</sub>), 46.3 (t, J= 22.0 Hz, CH), 43.0 (d, J = 3.8 Hz, CH<sub>2</sub>), 38.1 (CH<sub>2</sub>), 31.7 (d, J = 5.5 Hz, CH<sub>2</sub>), 29.1 (CH<sub>2</sub>), 22.7 (CH<sub>2</sub>), 14.1 (CH<sub>3</sub>). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>):  $\delta$  -113.12 (dd, J = 222.3, 24.4 Hz, 1F), -120.03 (d, J = 222.3 Hz, 1F). IR (neat):  $\nu_{\text{max}}$  1642w, 1496m, 1455m, 1220s, 760m, 700s cm<sup>-1</sup>. MS: m/z (%) relative intensity 382 (M<sup>+</sup> + 1, 14), 340 (100), 324 (9), 284 (7), 117 (12), 91 (16). HRMS (ESI-TOF) Calcd for  $C_{25}H_{30}F_2N [M + H]^+$ , 382.2348; found, 382.2322.

**Preparation of (2***R***\*,5***S***\*,9***bR***\*)-9***b***-Allyl-2-benzyl-1,1-diffuoro-5-isopropyl-2,3,5,9***b***-tetrahydro-1***H***-pyrrolo[2,1-***a***]isoindole (14).** *i***-Propylmagnesium chloride (0.3 mL, 0.6 mmol, 2 M solution in THF) was slowly added to a solution of 9a (0.04 g, 0.12 mmol) in THF (1 mL) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 4 h. The reaction mixture was cooled to 0 °C and 4 M HCl-dioxane in MeOH (4 M HCl-dioxane/MeOH, 1:29, 5 mL) was added, followed by the addition of excess NaBH<sub>3</sub>CN. After removal of the ice-bath, the reaction mixture was stirred for an additional 1 h and 10% aqueous NaOH solution (5 mL) was then added. The mixture was diluted with CH<sub>2</sub>Cl<sub>2</sub> and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 20 mL). The combined extracts were**  washed with brine and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Filtration followed by evaporation gave a crude product, which was purified by radial chromatography (SiO<sub>2</sub>, hexanes then 5% EtOAc in hexanes) to afford a pale yellow solid of 14 (0.029 g, 65% yield, mp = 87-89 °C). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.31–7.14 (m, 9H, ArH), 5.88-5.80 (m, 1H, CH), 5.06 (d, J = 17.2 Hz, 1H, CHH), 5.02 (d, J = 10.1 Hz, 1H, CHH), 3.79 (d, J = 4.7 Hz, 1H, CH), 3.41 (dd, J = 9.3, 7.7 Hz, 1H, CHH), 3.03 (dd, J = 13.8, 4.7 Hz, 1H, CHH), 3.03–2.93 (m, 1H, CH), 2.65–2.59 (m, 2H, CHH, CHH), 2.55 (t, *J* = 9.9 Hz, 1H, CHH), 2.49–2.45 (m, 1H, CHH), 1.97-1.90 (m, 1H, CH(CH<sub>3</sub>)<sub>2</sub>), 1.04 (d, J = 6.8 Hz, 3H, CH<sub>3</sub>), 0.76 (d, J = 6.7 Hz, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$ 143.3 (C), 139.4 (d, J = 6.8 Hz, C), 139.3 (C), 134.3 (CH), 128.7  $(2 \times CH)$ , 128.5  $(2 \times CH)$ , 127.8 (CH), 127.0 (dd, J = 264.0, 251.8 Hz, CF<sub>2</sub>), 126.9 (CH), 126.3 (CH), 122.8 (CH), 122.7 (CH), 118.0 (CH<sub>2</sub>), 80.0 (t, J = 22.0 Hz, C), 79.8 (CH), 59.2 (d, J = 9.8 Hz, CH<sub>2</sub>), 46.5 (t, J = 21.1 Hz, CH), 43.0 (d, J = 4.3 Hz, CH<sub>2</sub>), 34.0 (CH), 31.9 (d, J = 5.9 Hz, CH<sub>2</sub>), 20.6 (CH<sub>3</sub>), 18.0 (CH<sub>3</sub>). <sup>19</sup>F NMR (470 MHz, CDCl<sub>3</sub>):  $\delta$  –112.35 (dd, J = 222.8, 24.9 Hz, 1F), -120.23 (d, J = 222.3 Hz, 1F). IR (KBr):  $\nu_{\text{max}}$  3074br, 1458m, 1210s, 1017s, 761s, 699s cm<sup>-1</sup>. MS: m/z (%) relative intensity 368 (M<sup>+</sup> + 1, 7), 326 (100), 284 (26), 117 (14), 91 (14). HRMS (ESI-TOF) Calcd for C<sub>24</sub>H<sub>28</sub>F<sub>2</sub>N [M + H]<sup>+</sup>, 368.2192; found, 368.2243.

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**Supporting Information Available:** Experimental details and characterization data for all compounds and CIF data for singlecrystal X-ray analyses of *trans*-**5a** and *trans*-**6a**. This material is available free of charge via the Internet at http://pubs.acs.org.

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