

Fluoride-Catalyzed Addition of PhSCF2SiMe3 to *N***-Substituted Cyclic Imides Followed by Radical Cyclization: General Synthetic Strategy of** *gem***-Difluoromethylenated 1-Azabicyclic Compounds**

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*Recei*V*ed December 23, 2008*

PhSCF₂SiMe₃ (1) was found, for the first time, to undergo fluoride-catalyzed nucleophilic difluoro-(phenylsulfanyl)methylation reaction to cyclic imides **2**, affording the corresponding adducts **3** in moderate to good yields. Reductive cleavage of the phenylsulfanyl group of *N*-alkylated adducts 3 with Bu₃SnH/ AIBN yielded *gem*-difluoromethylated products **4**. Under the same reduction conditions, *N*-alkenylated and *N*-alkynylated adducts **3** afforded the corresponding *gem*-difluoromethylenated 1-azabicyclic compounds **5** and **6** with *trans* stereoselectivity. These compounds were employed as precursors for preparing substituted *gem*-difluoromethylenated pyrrolizidinones and indolizidinones **7** and **8** by treatment with Et₃SiH/BF₃ · OEt₂, and compounds **9** and **10** by nucleophilic displacement of the hydroxyl group, using organosilanes in the presence of BF₃·OEt₂. The synthesis of highly substituted *gem*-difluoromethylenated pyrrolizidines **13** and **14** was also demonstrated.

Introduction

gem-Difluoromethylene moiety has been incorporated into bioactive compounds due to its properties as an isopolar and isosteric substituent for an oxygen.¹ The presence of the *gem*difluoromethylene group can enhance the biological properties of drug molecules.2 Therefore, syntheses of *gem*-difluoromethylenated analogs of natural products have been extensively investigated.³ Few general synthetic methods for constructing

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gem-difluoromethylenated compounds were previously reported.4 It is still desirable to develop new flexible and facile strategies for the preparation of these compounds. Recently, PhSCF2SiMe3 (**1**) has been demonstrated as a useful *gem*difluoromethylene building block.⁵ Compound 1 reacts extensively with carbonyl derivatives,⁶ γ-ketoesters,⁷ imines,⁸ and alkyl bromides,⁹ providing convenient routes for the preparation of the corresponding *gem*-difluoromethylated alcohols, *γ*-lactones, amines, and alkanes, respectively. Furthermore, the stereoselective difluoromethylenation of chiral imines using PhSCF₂SiMe₃ (1) and the synthesis of chiral disubstituted *gem*difluoromethylenated pyrrolidines have been recently reported.10a In addition, alternative new entries to assemble various fluorinated *S*-, *O*-, and *N*-heterocycles have also been developed.10b-^d

As a part of our ongoing research interest in developing methodologies for the preparation of 1-azabicyclic derivatives, 11 we sought for a general method to synthesize *gem*-difluoromethylenated 1-azabicyclic analogs, employing **1** as a *gem*difluoromethylene building block. It has been demonstrated that cyclic imides can undergo fluoride-catalyzed trifluoromethylation, using (trifluoromethyl)trimethylsilane (CF_3SiMe_3) .¹² Hence, it is anticipated that fluoride-catalyzed nucleophilic difluoro-

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SCHEME 1. Fluoride-Catalyzed Nucleophilic Difluoro(phenylsulfanyl)methylation of 1 with Imides 2 and Synthetic Conversion of the Resulting Adducts 3 into Compounds 4, 5, and 6 and Subsequently to Compounds 13 and 14

(phenylsulfanyl)methylation of **1** to cyclic imides **2** would generate the corresponding adducts **3**. When R groups of adducts **3** are allylic and homoallylic, reductive cleavage of the corresponding phenylsulfanyl moiety could afford radical intermediate that should then undergo intramolecular radical cyclization, providing *gem*-difluoromethylenated 1-azabicyclic compounds **5** and **6**, respectively (Scheme 1). Furthermore, nucleophilic displacement of the hydroxyl group and subsequent organometallic addition to carbonyl group followed by reduction would lead to highly substituted difluoromethylenated 1-azabicyclic compounds **13** and **14**. The proposed synthetic operations are summarized in Scheme 1.

Results and Discussion

The initial investigation involved fluoride-catalyzed nucleophilic difluoro(phenylsulfanyl)methylation of imide **2a** with PhSCF₂SiMe₃ (1). The expected adduct **3a** was obtained in 95% yield when the reaction was carried out using 10 mol % of anhydrous tetrabutylammonium fluoride (TBAF) in dry THF at -78 °C to room temperature overnight followed by acidic workup (2 M HCl; Table 1, entry 1). To demonstrate the generality of the reaction, the reactions of **1** with *N*-substituted phthalimide and succinimide derivatives, which were prepared by base-catalyzed *N*-alkylation of imides¹³ with alkylating agents (Table 1, entries $1-6$, 8, $10-13$, 15 and 17), Mitsunobu reaction (Table 1, entries $1-6$, 8, $10-13$, 15 and 17), Mitsunobu reaction of imides with alcohols¹⁴ (Table 1, entries 7, 14 and 16), and cross-olefin metathesis¹⁵ of *N*-alkenylated imides with alkenes (Table 1, entries 9 and 18), were investigated. The results are summarized in Table 1. Except for some cases, the reactions of **1** with imide derivatives conducted under the standard conditions as for **3a** readily gave the corresponding adducts; yields range from moderate to good (Table 1). In case of imide **2i**, under the standard conditions, adduct with double bond transposition was obtained as a major product (33% yield). Gratifyingly, the reaction employing tetrabutylammonium triphenyldifluorosilicate (TBAT) in place of TBAF gave adduct **3i** in moderate yield

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a Isolated yields. *b* Unless otherwise noted, all reactions were carried out employing 10 mol % TBAF/THF, -78 °C to rt. ^{*c*} By using 10 mol % TBAT/DMF, -78 °C to rt. ^{*d*} By using 10 mol % TBAT/THF, -78 to 15 °C, 5 h. ^{*e*} By using 10 mol % TBAT/THF, -78 °C to rt. ^{*f*} Ratio of *E*- and
Zisomers was determined by ¹H NMR integration of the crude product. ^{*8} Z*-isomers was determined by ¹ H NMR integration of the crude product. *^g* Ratios of isomers (*trans*/*cis*) were determined by ¹ H NMR integration of the crude products.

(55%) and the double bond transposition adduct was not observed (Table 1, entry 9). It is also worth mentioning that the reaction of **1** with *N*-trimethylsilylpropargyl succinimide **2n** under the standard conditions furnished the expected product **3n** in only 2% yield, but the desilylated adduct **3o** was obtained as a major product (30% yield). However, the reaction employing 10 mol % of TBAT in dry DMF at -78 °C to room temperature gave the desired product **3n** in 45% yield together with **3o** (10% yield; Table 1, entry 14). Similar results, that is, **3n** (40% yield) and **3o** (8% yield) were obtained when the reaction was carried out using THF as the solvent. Interestingly, treatment of **1** with *N*-propargylphthalimide **2m** employing 10 mol % TBAT in DMF at -78 °C to room temperature unexpectedly provided trimethylsilylated derivative **3m** in 68% yield (Table 1, entry 13).

The reductive desulfenylation of adducts **3a**-**^e** was achieved by treating with Bu3SnH and a catalytic amount of AIBN in refluxing toluene. The *gem*-difluoromethylated products **4a**-**^e** were obtained in moderate to good yields (66-86%; Table 1,

SCHEME 2. Proposed Transition States for the Radical Cyclization

Bu₃SnH, AIBN A (favored TS) $3₁$ trans-5a cis-5a (minor (major) Bu₃SnH, AIBN Ċ C (favored TS) 3_D 'n trans-6a cis-6a

(major)

entries $1-5$).^{6b,16} The *gem*-difluoromethyl radical intermediate, generated by CF_2 -SPh bond cleavage of the adducts **3** containing the *N*-unsaturated substituents, could be trapped intramolecularly by an unsaturated functional moiety,17 affording *gem*difluoromethylenated 1-azabicyclic compounds **5**. Thus, under similar radical conditions, intramolecular radical cyclization of **3f** readily proceeded to provide **5a** in 73% yield as a 94:6 mixture of *trans*- to *cis*-isomers together with a small amount of the corresponding reduced product. A single *trans*-**5a** was obtained in pure form after purification by either preparative thin-layer chromatography on silica gel or crystallization. Its relative stereochemistry was established by X-ray crystallography (see Supporting Information).

To demonstrate the efficiency and viability of our methodology, series of *gem*-difluoromethylenated pyrrolizidine derivatives **5b**-**g**, as mixtures of isomers, were synthesized, in moderate to good yields, by treatment of adducts **3g**-**^l** under standard radical conditions (Table 1, entries $7-12$). Based on the X-ray crystallographic data of the major *trans*-isomer of **5a**, we assumed that the *trans*-isomers of **5b**-**^g** were the major isomers. Notably, similar radical cyclization of *N*-propargyl substituted adducts proceeded smoothly, providing the cyclized products in moderate yields (Table 1, entries 13-15). Cyclization of silylated alkynyl derivatives **3m** and **3n** afforded compounds **5h** and **5i** as a mixture of *E*- and *Z*-isomers. The *E*-isomer was obtained as a major isomer, of which the stereochemistry was established by NOE experiments (see Supporting Information). Irradiation of the olefinic proton of **5h** or **5i** led to no enhancement of the allylic protons. The formation of tributylstannylated product **5j** (Table 1, entry 15) presumably resulted from consecutive addition of the tributylstannyl group to the triple bond followed by *gem*-difluoromethyl radical addition to the resulting vinylstannane derivative.18 The presence of the exocyclic methylene moiety in compounds of types **5h** and **5i** should be valuable for further synthetic modification.

(minor)

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Having established an efficient access to *gem*-difluoromethylenated pyrrolizidine derivatives **5**, we further demonstrated that our method could be used as a general route for the synthesis of *gem*-difluoromethylenated indolizidine derivatives **⁶**. Thus, *^N*-homoallylated adducts **3p**-**^r** were treated with Bu3SnH/AIBN in refluxing toluene, yielding the corresponding *gem*-difluoromethylenated indolizidines **6a**-**^c** as a mixture of *trans*- and *cis*-isomers (Table 1, entries 16-18). The relative stereochemistry of the major *trans*-**6a** was confirmed by X-ray crystallography (see Supporting Information).

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SCHEME 3. Lewis Acid-Catalyzed Nucleophilic Substitution of the Hydroxyl Group of Compounds 5b and 6a

The stereochemical outcomes of compounds **5** and **6** can be rationalized as shown in Scheme 2. The radical mediated cyclization was proposed to proceed via 5-*exo* or 6-*exo* cyclization mode.19 Transition states **A** and **C** are energetically more favorable due to minimized steric interaction between the pseudoaxial hydroxyl group and the vinylic double bond, resulting in the formation of thermodynamically more stable *trans*-isomer of pyrrolidine or piperidine heterocycles.

Having a general route to synthesize derivatives **5** and **6**, we subsequently focused on the synthetic utilities of these adducts as precursors for the preparation of substituted *gem*-difluoromethylenated pyrrolizidines and indolizidines. It is anticipated that the presence of hydroxyl group in adducts **5** and **6** should provide a convenient access to an iminium intermediate, $20,21$ which, in principle, can be trapped by an appropriate nucleophile. The results are summarized in Scheme 3. The adducts **5b** (*trans*/*cis*; 77:23) and **6a** (*trans*/*cis*; 88:12) were reduced with BF_3 OEt₂ and Et₃SiH²² in CH₂Cl₂ at -78 °C to room temperature, generating the corresponding lactams **7** (77% yield) and **8** (98% yield) as a single isomer, as confirmed by the NOE experiments (see Supporting Information). Alkylation of **5b** with Grignard reagents (MeMgBr or *i*-PrMgCl in THF in the presence of BF_3 OEt₂ at 0 °C to rt, overnight) or organocoppers $(n-Bu_2CuLi$ or $n-Bu_2Cu(CN)Li_2$ in THF in the presence of BF_3 · OEt₂ at -30 °C to rt overnight) was unsuccessful, and only

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SCHEME 4. Proposed Transition States for the Formation

starting material was recovered. However, allylation reaction of **5b**, employing allyltrimethylsilane with $BF_3 \cdot OEt_2$ in CH_2Cl_2 at -78 °C to rt overnight, led to **9a** in low yield (10% yield) together with recovery of **5b**. Gratifyingly, when the reaction was performed under refluxing conditions for 3 h, **9a** was obtained in moderate yield (67% yield). Similarly, under the same conditions, **10a** was synthesized in 53% yield from the corresponding compound **6a**. Both compounds **9a** and **10a** were obtained as a single isomer (see Supporting Information). Similarly, treatment of **5b** and **6a** with cinnamyltrimethylsilane afforded adducts **9b** and **10b**, respectively, as a mixture of diastereomers, as shown in Scheme 3. The reactions of **5b** and **6a** with phenyltrimethylsilylacetylene employing trimethylsilyl triflate (TMSOTf) as the Lewis acid at -78 °C to rt overnight provided the respective adducts **9c** (54% yield as a 93:7 mixture of isomers) and **10c** (67% yield as a 86:14 mixture of isomers). The use of BF_3 · OEt₂, SnCl₄, and TiCl₄ did not furnish to the desired products but led to the recovery of the starting material **5b** and phenylacetylene. This result may be due to extensive decomposition of phenyltrimethylsilylacetylene to phenylacetylene prior to its addition to iminium ion intermediate. The relative *trans* stereochemistries of adducts **7** and **9** could be rationalized based on the fact that the nucleophiles (hydride or organosilanes) attack the initially formed iminium ion intermediate from the face opposite to the benzyl group to avoid steric interaction (Scheme 4). In addition, attack of nucleophiles to the iminium ion derived from **6a** from the pseudoaxial direction

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SCHEME 5. Preparation of Alkyl-Substituted *gem***-Difluoromethylenated Pyrrolizidines 11, 13, and 14 and Indolizidine 12**

could be further explained by stereoelectronic principle, leading preferably to the *trans*-isomers of adducts **8** and **10**.

The conversions of compounds **9a** and **10a** to alkylsubstituted *gem*-difluoromethylenated pyrrolizidines and indolizidines proceeded smoothly (Scheme 5). Reduction of compounds 9a and 10a using LiAlH₄ in refluxing THF provided the corresponding *gem*-difluoromethylenated pyrrolizidine **11** (84% yield) and indolizidine **12** (60% yield), respectively (Scheme 5). The reaction of **9a** with *n*-BuLi/CeCl₃ in THF²³ followed by acidification generating in situ iminium ion intermediate, which was subsequently reduced by sodium cyanoborohydride (NaBH3CN) to afford butyl-substituted *gem*difluoromethylenated pyrrolizidine **13** in 80% yield as a single diastereomer. Similarly, a single diastereomer of isopropylsubstituted *gem*-difluoromethylenated pyrrolizidine **14** was produced in 65% yield. The relative stereochemistries of compounds **¹¹**-**¹⁴** were established by NOE experiments (see Supporting Information). The observed stereochemical outcomes of the reactions can be explained that the addition of organometallic reagents followed by acidification afforded iminium ions which were preferably reduced from the less sterically hindered face opposite to the allyl substituent (Scheme 5). It is worth mentioning that compound **10a** did not react with either *n*-BuLi/CeCl₃ or *i*-PrMgCl in THF under the same conditions as for **9a**.

Conclusion

In conclusion, we have successfully developed for the novel fluoride-catalyzed nucleophilic difluoro(phenylsulfanyl)methylation reactions of PhSCF₂SiMe₃ to succinic and phthalimide derivatives. The resulting adducts were employed as useful precursors for the preparation of *gem*-difluoromethylenated pyrrolizidines and indolizidines. The synthetic operation involves sequential intramolecular radical cyclization, nucleophilic displacement of the hydroxyl functionality using Lewis acid/ allyltrimethylsilanes or trimethylsilylacetylenes, organometallic addition employing *n*-BuLi/CeCl₃ and *i*-PrMgCl onto the carbonyl group of the *γ*-lactam moiety, followed by reduction of the resulting adducts with NaBH3CN. The synthetic application of our method for preparation of *gem*-difluoromethylenated analogs of some polyhydroxylated natural products is currently being investigated.

IOC Article

Experimental Section

General Procedure for the Preparation of Compounds 3. 2-Benzyl-3-(difluoro(phenylsulfanyl)methyl)-3-hydroxy-isoindolin-1-one (3a). To a mixture of compound **1** (0.928 g, 4 mmol) and 2-benzylisoindoline-1,3-dione (**2a**; 0.470 g, 2 mmol) in THF (5 mL), was added 10 mol % TBAF (0.4 mL, 0.4 mmol, 1 M solution in THF). The reaction mixture was stirred at -78 °C followed by slowly warming up to room temperature overnight. The solution was quenched with 1 M HCl (3 mL) and extracted with EtOAc (3 \times 25 mL). The organic phase was washed successively with water and brine and dried over anhydrous $Na₂SO₄$. After solvent removal, the crude product was purified by radial chromatography $(SiO₂,$ ¹⁰-20% EtOAc in hexanes) to give a white crystal of **3a** (0.753 g, 95% yield, mp = 142-144 °C). ¹H NMR (300 MHz, CDCl₃):
 δ 7 86-7 83 (m 1H ArH) 7 72-7 68 (m 1H ArH) 7 60-7 58 *^δ* 7.86-7.83 (m, 1H, Ar*H*), 7.72-7.68 (m, 1H, Ar*H*), 7.60-7.58 (m, 2H, Ar*H*), 7.35-7.25 (m, 3H, Ar*H*), 7.20-7.10 (m, 7H, Ar*H*), 4.95 (d, $J = 15.6$ Hz, 1H, CHH), 4.45 (d, $J = 15.6$ Hz, 1H, CHH), 3.85 (br s, 1H, OH). ¹³C NMR (125 MHz, CDCl₃): δ 168.3 (C=O), 141.9 (C), 137.5 (2 × C), 136.7 (2 × CH), 132.8 (CH), 131.5 (C), 131.0 (2 \times CH), 130.1 (CH), 129.24 (t, *J* = 187.5 Hz, CF₂), 129.0 $(2 \times CH)$, 128.5 (3 \times CH), 127.4 (CH), 124.2 (CH), 123.8 (CH), 91.2 (t, $J = 25.0$ Hz, C), 43.5 (CH₂). ¹⁹F NMR (470 MHz, CDCl₃): *δ* -80.48 (d, *J* = 211.0 Hz, 1F), -81.57 (d, *J* = 211.0 Hz, 1F). IR (nujol): *ν*max 3207br, 1682s, 1612s, 1612m, 1497m, 1471s, 1456s, 1455s, 1427m, 1397s, 1361s, 1163s, 1137m, 1108m, 1073m, 1061s, 1027m, 1012m, 974m, 945m, 907m, 897m, 886m, 845m, 823m, 770s, 752s, 709s, 700s, 690s cm-¹ . MS: *m*/*z* (%) relative intensity 398 (M+, 10), 380 (7), 250 (6), 238 (47), 161 (11), 160 (100), 91 (20), 65 (7). HRMS (ESI-TOF) Calcd for $C_{22}H_{17}F_2NO_2SNa$ [M + Na]⁺, 420.0846; found, 420.0851.

General Procedure for the Reductive Cyclization of Compounds 3 to Compounds 5. Preparation of 1,1-Difluoro-9bhydroxy-2-methyl-2,3-dihydro-1*H***-pyrrolo[2,1-a]isoindol-5(9b***H***) one (5a).** Argon was bubbled through a solution of **3f** (0.374 g, 1 mmol) in toluene (2 mL) for 30 min, and a mixture of Bu₃SnH (0.47 mL, 1.75 mmol) and AIBN (25 mg, 0.15 mmol) in toluene

⁽²³⁾ Nukui, S.; Sodeoka, M.; Sasai, H.; Shibasaki, M. *J. Org. Chem.* **1995**, *60*, 398–404.

(8 mL) was added dropwise at reflux overa1h period followed by refluxing for an additional 4 h. Volatiles were evaporated and the tin byproduct were removed by column chromatography $(SiO₂,$ $CH₂Cl₂$ then EtOAc) to give a crude product, which was purified by column chromatography $(SiO₂, 20\%$ EtOAc in hexanes) to afford white crystals of **5a** (0.191 g, 73% yield, mp = $122-125$ °C) as a 94:6 mixture of *trans*- and *cis*-isomers. ¹H NMR (500 MHz, CDCl₃, *cis*-isomer marked*): *^δ* 7.68-7.64 (m, 6H, ArH of *trans*- and *cis*isomers), 7.57-7.55 (m, 2H, ArH of *trans*- and *cis*-isomers), 4.12* (dd, $J = 12.0$, 9.3 Hz, 1H, CHH), 3.72 (br s, 2H, OH of *trans*- and *cis*-isomers), 3.68-3.63 (m, 1H, CHH), 3.39-3.27 (m, 3H, CHH of the *trans*-isomer and CH of *trans*- and *cis*-isomers), 3.13* (dd, *J* = 12.0, 4.9 Hz, 1H, CHH), 1.50* (d, *J* = 7.5 Hz, 3H, CH₃), 1.19 (d, *^J*) 6.8 Hz, 3H, CH3). 13C NMR (125 MHz, CDCl3): *^δ* 170.7 (C=O), 141.4 (C), 133.2 (CH), 132.9 (C), 130.9 (CH), 123.9 (CH), 123.2 (CH), 122.9 (t, $J = 258.2$ Hz, CF₂), 93.1 (t, $J = 25.1$ Hz, C), 46.4 (d, $J = 7.4$ Hz, CH₂), 39.3 (t, $J = 22.7$ Hz, CH), 9.5 (d, $J = 7.1$ Hz, CH₃). ¹⁹F NMR (470 MHz, CDCl₃, *cis*-isomer marked*): δ -101.16* (dd, *J* = 232.2, 23.5 Hz, 1F), -126.86* (d, *J* = 232.2 Hz, 1F), -127.82 (dd, *J* = 220.9, 24.9 Hz, 1F), -129.60 (dd, *J* = 220.9, 6.8 Hz, 1F). IR (nujol): *ν*_{max} 3237br, 1681s, 1616s, 1469s, 1378m, 1239m, 1147m, 1123m, 1958m, 1003m, 759s, 709m, 701m, 604m cm-¹ . MS: *m*/*z* (%) relative intensity 239 (M+, 33), 203 (22), 201 (62), 200 (100), 188 (66), 172 (23), 161 (34), 160 (42), 145 (27), 133 (48), 132 (22), 125 (25), 117 (59), 91 (28), 77 (41). HRMS (ESI-TOF) Calcd for $C_{12}H_{11}F_2NO_2Na$ [M + Na]⁺, 262.0656; found, 262.0634.

General Procedure for the Preparation of Compounds 9 and 10. Preparation of (2*R****,9b***R****)-9b-Allyl-2-benzyl-1,1-difluoro-2,3** dihydro-1*H*-pyrrolo[2,1-*a*]isoindol-5(9b*H*)-one (9a). $BF_3 \cdot OEt_2$ (0.6 mL, 4.93 mmol) and allyltrimethylsilane (2.6 mL, 16.5 mmol) were added to a stirred solution of **5b** (1.04 g, 3.3 mmol) in CH_2Cl_2 (12 mL) at room temperature and the mixture was heated at reflux for 3 h. The reaction was quenched with a saturated aqueous sodium hydrogen carbonate and extracted with CH_2Cl_2 (3 \times 20 mL). The combined extracts were washed with brine and dried over anhydrous Na2SO4. Filtration followed by evaporation gave a crude product, which was purified by preparative thin-layer chromatography $(SiO₂,$ 10% EtOAc in hexanes) to give a white solid of **9a** as a *trans*isomer (0.75 g, 67% yield, mp = 91–93 °C). ¹H NMR (500 MHz,
CDCl₂): δ 7.83 (d, *I* = 7.6 Hz, 1H, ArH) 7.63 (td, *I* = 7.5, 1.2 CDCl₃): δ 7.83 (d, *J* = 7.6 Hz, 1H, ArH), 7.63 (td, *J* = 7.5, 1.2 Hz, 1H, ArH), 7.54 (td, $J = 7.5$, 1.0 Hz, 1H, ArH), 7.50 (d, $J =$ 7.6 Hz, 1H, ArH), 7.35 (t, $J = 7.6$ Hz, 2H, ArH), 7.30-7.25 (m, 3H, ArH), 5.31 (ddt, *J* = 17.1, 10.1, 7.1 Hz, 1H, CH), 5.03 (dd, *J*) 17.1, 1.4 Hz, 1H, CHH), 4.97 (d, *^J*) 10.0 Hz, 1H, CHH), 3.73 (dd, $J = 12.1$, 8.7 Hz, 1H, CHH), 3.56 (dd, $J = 12.1$, 9.5 Hz, 1H, CHH), 3.39–3.29 (m, 1H, CH), 3.11 (dd, $J = 13.9, 5.3$ Hz, 1H, CHH), 2.91 (dd, *J* = 14.1, 7.0 Hz, 1H, CHH), 2.75 (dd, *J* = 13.9, 9.4 Hz, 2H, CHH). ¹³C NMR (125 MHz, CDCl₃): δ 172.1 (C=O), 142.8 (d, J = 4.3 Hz, C), 137.9 (C), 133.8 (C), 132.4 (CH), 129.7 (CH), 129.4 (CH), 128.74 (2 × CH), 128.67 (2 × CH), 126.8 (CH), 124.9 (dd, *J* = 263.4, 255.4 Hz, CF₂), 124.2 (CH), 122.5 (CH), 120.2 (CH₂), 74.5 (t, $J = 25.3$ Hz, C), 48.3 (t, $J = 22.6$ Hz, CH), 45.1 (d, $J = 7.4$ Hz, CH₂), 37.0 (d, $J = 2.4$ Hz, CH₂), 32.9 (d, $J =$ 7.8 Hz, CH₂). ¹⁹F NMR (470 MHz, CDCl₃): δ -118.14 (dd, *J* = 226.5, 23.0 Hz, 1F), -121.05 (dd, $J = 226.5, 7.1$ Hz, 1F). IR (KBr): *ν*max 1703s, 1643w, 1612w, 1491w, 1467m, 1363s, 1220s, 1049s, 751m, 704s cm⁻¹. MS: m/z (%) relative intensity 340 (M⁺ + 1, 11) 339 (M⁺ 2) 298 (100) 249 (20) 91 (64) 65 (11) HRMS 11), 339 (M+, 2), 298 (100), 249 (20), 91 (64), 65 (11). HRMS (ESI-TOF) Calcd for $C_{21}H_{19}F_2NONa [M + Na]⁺$, 362.1332; found, 362.1319.

Preparation of (2*R****,9b***R****)-9b-Allyl-2-benzyl-1,1-difluoro-2,3,5,9btetrahydro-1***H***-pyrrolo[2,1-***a***]isoindole (11).** To a suspension of LiAlH4 (4.3 mg, 1.13 mmol) in THF (3 mL) was added a solution of **9a** (0.153 g, 0.45 mmol) in THF (4 mL). The mixture was heated at reflux overnight and quenched at 0 °C by careful addition of water (0.5 mL) followed by 1 M NaOH (0.5 mL). The resulting mixture was filtered through a Celite pad and the filtrate was concentrated under reduced pressure. Purification of the crude

product by column chromatography ($SiO₂$, $5%$ EtOAc in hexanes) gave a pale yellow oil of 11 (0.123 g, 84% yield). ¹H NMR (500 MHz, CDCl3): *^δ* 7.30-7.25 (m, 5H, ArH), 7.21-7.15 (m, 4H, ArH), 5.50 (ddt, *J* = 17.2, 10.1, 7.0 Hz, 1H, CH), 4.95 (dd, *J* = 17.2, 1.8 Hz, 1H, CHH), 4.91 (d, $J = 10.2$ Hz, 1H, CHH), 4.23 (d, *J* = 15.1 Hz, 1H, CHH), 3.76 (d, *J* = 15.1 Hz, 1H, CHH), 3.29 (t, *J* = 7.4 Hz, 1H, CHH), 3.02 (dd, *J* = 13.9, 5.1 Hz, 1H, CHH), 2.90-2.86 (m, 1H, CH), 2.63-2.58 (m, 2H, CHH, CHH), 2.46 (dd, $J = 13.9, 7.4$ Hz, 1H, CHH), 2.42 (ddd, $J = 11.2, 9.1, 1.7$ Hz, 1H, CHH). ¹³C NMR (125 MHz, CDCl₃): δ 140.7 (C), 139.1 (C), 138.8 (d, $J = 7.5$ Hz, C), 132.8 (d, $J = 1.4$ Hz, CH), 128.6 (2 \times CH), 128.5 (2 \times CH), 127.9 (CH), 127.0 (CH), 126.9 (t, *J* = 261.0 Hz, CF₂), 126.3 (CH), 123.9 (CH), 122.5 (CH), 118.4 (CH₂), 81.5 $(t, J = 22.8 \text{ Hz}, \text{C}), 60.2 \text{ (CH}_2), 57.3 \text{ (d, } J = 9.1 \text{ Hz}, \text{CH}_2), 46.6 \text{ (t,)}$ $J = 21.6$ Hz, CH), 43.1 (t, $J = 3.0$ Hz, CH₂), 31.0 (d, $J = 6.0$ Hz, CH₂). ¹⁹F NMR (470 MHz, CDCl₃): δ -109.23 (dd, *J* = 224.2, 23.0 Hz, 1F), -115.08 (dd, $J = 223.7$, 6.1 Hz, 1F). IR (neat): v_{max} 1641m, 1456s, 1218s, 724s cm-¹ . MS: *m*/*z* (%) relative intensity 278 (9), 236 (100), 216 (7), 194 (18), 176 (7), 130 (6). HRMS (ESI-TOF) Calcd for $C_{21}H_{22}F_2N$ [M + H]⁺, 326.1722; found, 326.1767.

Preparation of (2*R****,5***S****,9b***R****)-9b-Allyl-2-benzyl-5-butyl-1,1 difluoro-2,3,5,9b-tetrahydro-1***H***-pyrrolo[2,1-***a***]isoindole (13).** To a solution of CeCl₃ (0.65 g, 1.75 mmol) in THF (5 mL) was added n -BuLi (1.2 mL, 1.74 mmol, 1.46 M solution in hexane) at -78 °C. After stirring at -⁷⁸ °C for 1 h, a solution of **9a** (0.12 g, 0.35 mmol) in THF (5 mL) was added. The mixture was stirred at -78 °C for 10 h and then gradually warmed to -20 °C, and quenched by the addition of 4 M HCl-dioxane in MeOH (4 M HCl-dioxane/ MeOH, 1:29, 5 mL) followed by excess NaBH₃CN. The resulting reaction mixture was stirred at 0 °C for 1 h and 10% aqueous NaOH solution (5 mL) was then added. The mixture was diluted with CH_2Cl_2 and extracted with CH_2Cl_2 (3 \times 20 mL). The combined extracts were washed with brine and dried over anhydrous $Na₂SO₄$. Filtration followed by evaporation gave a crude product, which was purified by radial chromatography ($SiO₂$, 5% EtOAc in hexanes) to afford a colorless oil of **13** (0.094 g, 80% yield). ¹ H NMR (500 MHz, CDCl3): *^δ* 7.31-7.18 (m, 8H, ArH), 7.13-7.11 (m, 1H, ArH), 5.59 (ddt, $J = 17.3$, 10.4, 7.1 Hz, 1H, CH), 4.96-4.92 (m, 2H, CH₂), 3.83 (dd, $J = 9.1$, 3.4 Hz, 1H, CH), 3.40 (t, $J = 7.9$ Hz, 1H, CHH), 3.04 (dd, $J = 13.7, 4.9$ Hz, 1H, CHH), 2.97-2.90 (m, 1H, CH), 2.63 (dd, $J = 13.7$, 9.7 Hz, 1H, CHH), 2.57-2.53 (m, 2H, CHH, CHH), 2.47 (dd, $J = 13.9, 7.3$ Hz, 1H, CHH), 1.72-1.68 (m, 1H, CHH), $1.52-1.26$ (m, 5H, CHH, $2 \times$ CH₂), 0.89 (t, $J =$ 7.3 Hz, 3H, CH3). 13C NMR (125 MHz, CDCl3): *δ* 144.9 (C), 139.2 (C), 138.2 (d, *J* = 6.8 Hz, C), 133.9 (CH), 128.7 (2 × CH), 128.5 $(2 \times CH)$, 127.9 (CH), 127.0 (dd, $J = 263.5, 251.9$ Hz, CF₂), 127.0 (CH), 126.3 (CH), 123.0 (CH), 122.3 (CH), 118.0 (CH2), 80.5 (t, $J = 22.0$ Hz, C), 73.5 (CH), 58.4 (d, $J = 9.6$ Hz, CH₂), 46.3 (t, *J* $= 22.0$ Hz, CH), 43.0 (d, $J = 3.8$ Hz, CH₂), 38.1 (CH₂), 31.7 (d, $J = 5.5$ Hz, CH₂), 29.1 (CH₂), 22.7 (CH₂), 14.1 (CH₃). ¹⁹F NMR (470 MHz, CDCl₃): δ -113.12 (dd, $J = 222.3$, 24.4 Hz, 1F), -120.03 (d, *^J*) 222.3 Hz, 1F). IR (neat): *^ν*max 1642w, 1496m, 1455m, 1220s, 760m, 700s cm⁻¹. MS: m/z (%) relative intensity 382 (M⁺ + 1, 14), 340 (100), 324 (9), 284 (7), 117 (12), 91 (16). HRMS (ESI-TOF) Calcd for $C_{25}H_{30}F_{2}N$ [M + H]⁺, 382.2348; found, 382.2322.

Preparation of (2*R****,5***S****,9b***R****)-9b-Allyl-2-benzyl-1,1-difluoro-5-isopropyl-2,3,5,9b-tetrahydro-1***H***-pyrrolo[2,1-***a***]isoindole (14).** *i*-Propylmagnesium chloride (0.3 mL, 0.6 mmol, 2 M solution in THF) was slowly added to a solution of 9a (0.04 g, 0.12 mmol) in THF (1 mL) at 0 °C. The reaction mixture was warmed to room temperature and stirred for 4 h. The reaction mixture was cooled to 0 °C and 4 M HCl-dioxane in MeOH (4 M HCl-dioxane/MeOH, 1:29, 5 mL) was added, followed by the addition of excess NaBH₃CN. After removal of the ice-bath, the reaction mixture was stirred for an additional 1 h and 10% aqueous NaOH solution (5 mL) was then added. The mixture was diluted with CH_2Cl_2 and extracted with CH_2Cl_2 (3 \times 20 mL). The combined extracts were washed with brine and dried over anhydrous Na₂SO₄. Filtration followed by evaporation gave a crude product, which was purified by radial chromatography (SiO₂, hexanes then 5% EtOAc in hexanes) to afford a pale yellow solid of **14** (0.029 g, 65% yield, mp = 87-89 °C). ¹H NMR (500 MHz, CDCl₃): *δ* 7.31-7.14 (m,
9H ArH) 5.88-5.80 (m, 1H CH) 5.06 (d, *I* = 17.2 Hz, 1H 9H, ArH), 5.88-5.80 (m, 1H, CH), 5.06 (d, $J = 17.2$ Hz, 1H, CHH), 5.02 (d, $J = 10.1$ Hz, 1H, CHH), 3.79 (d, $J = 4.7$ Hz, 1H, CH), 3.41 (dd, $J = 9.3, 7.7$ Hz, 1H, CHH), 3.03 (dd, $J = 13.8, 4.7$ Hz, 1H, CHH), 3.03-2.93 (m, 1H, CH), 2.65-2.59 (m, 2H, CHH, CHH), 2.55 (t, $J = 9.9$ Hz, 1H, CHH), 2.49-2.45 (m, 1H, CHH), 1.97-1.90 (m, 1H, CH(CH₃)₂), 1.04 (d, $J = 6.8$ Hz, 3H, CH₃), 0.76 (d, *^J*) 6.7 Hz, 3H, CH3). 13C NMR (125 MHz, CDCl3): *^δ* 143.3 (C), 139.4 (d, *J* = 6.8 Hz, C), 139.3 (C), 134.3 (CH), 128.7 $(2 \times CH)$, 128.5 $(2 \times CH)$, 127.8 (CH), 127.0 (dd, $J = 264.0$, 251.8 Hz, CF₂), 126.9 (CH), 126.3 (CH), 122.8 (CH), 122.7 (CH), 118.0 (CH₂), 80.0 (t, $J = 22.0$ Hz, C), 79.8 (CH), 59.2 (d, $J = 9.8$ Hz, CH₂), 46.5 (t, $J = 21.1$ Hz, CH), 43.0 (d, $J = 4.3$ Hz, CH₂), 34.0 (CH), 31.9 (d, $J = 5.9$ Hz, CH₂), 20.6 (CH₃), 18.0 (CH₃). ¹⁹F NMR (470 MHz, CDCl₃): δ -112.35 (dd, *J* = 222.8, 24.9 Hz, 1F), -120.23 (d, *J* = 222.3 Hz, 1F). IR (KBr): *ν*_{max} 3074br, 1458m, 1210s, 1017s, 761s, 699s cm-¹ . MS: *m*/*z* (%) relative intensity 368 $(M^+ + 1, 7)$, 326 (100), 284 (26), 117 (14), 91 (14). HRMS (ESI-TOF) Calcd for $C_{24}H_{28}F_2N$ [M + H]⁺, 368.2192; found, 368.2243.

Acknowledgment. Financial support from the Center for Innovation in Chemistry (PERCH-CIC), the Commission on Higher Education (CHE-RES-RG), Ministry of Education, and the Thailand Research Fund (to M.P., BRG49800005; and to P.K., BRG50080008) is gratefully acknowledged.

Supporting Information Available: Experimental details and characterization data for all compounds and CIF data for singlecrystal X-ray analyses of *trans*-**5a** and *trans*-**6a**. This material is available free of charge via the Internet at http://pubs.acs.org.

JO802794U